Course Specifications

Valid as from the academic year 2015-2016

Mass Spectrometry & Isotopic Analysis (C002564)

Course size

<table>
<thead>
<tr>
<th>Credits</th>
<th>Study time</th>
<th>Contact hrs</th>
</tr>
</thead>
<tbody>
<tr>
<td>6.0</td>
<td>153 h</td>
<td>57.0 h</td>
</tr>
</tbody>
</table>

Course offerings and teaching methods in academic year 2017-2018

A (semester 2)
- seminar: coached exercises 12.5 h
- lecture 25.0 h
- practicum 10.0 h
- self-reliant study activities 12.5 h

Lecturers in academic year 2017-2018

Vanhaecke, Frank WE06 lecturer-in-charge
Boeckx, Pascal LA08 co-lecturer

Offered in the following programmes in 2017-2018

<table>
<thead>
<tr>
<th>programme</th>
<th>crdts</th>
<th>offering</th>
</tr>
</thead>
<tbody>
<tr>
<td>Master of Science in Chemistry</td>
<td>6</td>
<td>A</td>
</tr>
<tr>
<td>Master of Science in Chemical Engineering</td>
<td>6</td>
<td>A</td>
</tr>
<tr>
<td>Master of Science in Chemical Engineering</td>
<td>6</td>
<td>A</td>
</tr>
<tr>
<td>Exchange Programme in Chemistry (master's level)</td>
<td>6</td>
<td>A</td>
</tr>
</tbody>
</table>

Teaching languages

English

Keywords

Mass spectrometry, isotope ratios

Position of the course

In this course, the students first obtain insight into the sources of natural variation in the isotopic composition of the elements. Subsequently, they are familiarized with the basic principles of the most important types of mass spectrometers and their use in instruments deployed in the context isotopic analysis. The capabilities offered by isotopic analysis are illustrated via real-life application from various research fields. This course aims at providing the student with a quite profound insight into the techniques that are deployed in the context described above, their application areas and the capabilities and limitations of up-to-date instrumentation.

Contents

An introduction to inorganic mass spectrometry / definition of terms
Types of mass spectrometers (sector field, quadrupole-based, time-of-flight)
The isotopic composition of the elements
Natural abundance, fractionation and notation for light stable isotopes
ICP – mass spectrometry (ICP-MS)
Thermal ionization mass spectrometry (TIMS)
Multi-collector ICP – mass spectrometry (MC-ICP-MS)
Accelerator mass spectrometry (AMS, for 14C-dating & other applications)
Principles of measurement of light stable isotopes via isotope ratio mass spectrometry (IRMS)
Compound-specific stable isotope analyses via GC and HPLC-IRMS
Laser-based spectroscopy for light element stable isotope analysis
Isotope ratio applications: Sr, Pb, nuclear applications, stable isotopic tracer experiments, applications based on natural isotope fractionation effects, source identification and biodegradation of pollutants (e.g., nitrate and BTEX), greenhouse gas emission and source apportionment (nitrate, greenhouse gases), 15N tracing studies

(Approved)
for N biogeochemistry, stable isotope probing and use of biomarkers (e.g., microbial community structures, soil C dynamics and paleo-climatology)

**Initial competences**
Successful completion of the courses `Analytical chemistry: principles´ and `Spectroscopic methods of analysis´ or having mastered the corresponding competences in another way.

**Final competences**
1. Insight into the causes of natural variation in the isotopic composition of the elements.
2. Having an overview over the mass spectrometric methods that can be deployed for isotopic analysis, their application areas, capabilities and limitations.
3. Understanding and being able to explain the basic operating principles of analytical instrumentation for isotopic analysis.
4. Selecting an appropriate mass spectrometric technique for isotopic analysis for an analytical problem in this context.
5. Awareness of applications relying on isotopic analysis, including elemental assay via isotope dilution, dating, provenance determination, isotope ratios as paleoproxies and biomarkers, environmental studies.
6. Adequately interpreting a paper on isotopic analysis from the international literature.
7. Solving problems in the context of isotopic analysis via mass spectrometric techniques.

**Conditions for credit contract**
Access to this course unit via a credit contract is determined after successful competences assessment

**Conditions for exam contract**
This course unit cannot be taken via an exam contract

**Teaching methods**
Lecture, practicum, self-reliant study activities, seminar: coached exercises

**Extra information on the teaching methods**
Assignment – the student selects a paper out of a selection of publications from international peer-reviewed journals, assesses which technique was used, what the purpose of the investigation was, how the measurements were carried out, etc and communicates this to his/her fellow students via an oral presentation.

**Learning materials and price**
Syllabus in English. Estimated cost: 20.0 €

**References**

**Course content-related study coaching**
Answering of questions via e-mail or during a personal meeting after appointment, made by e-mail.

**Evaluation methods**
end-of-term evaluation and continuous assessment

**Examination methods in case of periodic evaluation during the first examination period**
Written examination with open questions, oral examination

**Examination methods in case of periodic evaluation during the second examination period**
Written examination with open questions

**Examination methods in case of permanent evaluation**
Assignment

**Possibilities of retake in case of permanent evaluation**
(Approved)
not applicable

**Extra information on the examination methods**

**Theory:**
Written examination with oral explanation.
The examination consists of overview questions, more detailed questions on specific
course subjects and questions aiming at assessing the student’s understanding of the
matter. Exercises are also included in the theoretical exam
Permanent evaluation via an assignment - the student studies a relevant publication
from an international peer-reviewed journal, assesses which technique was used, what
the purpose of the investigation was, how the measurements were carried out, etc and
communications this to his/her fellow students via an oral presentation. Students who are
absent without any well-justified reason or who do not participate in all evaluation
methods (practicals) of the continuous assessment, get a non-deliberable examination
mark. The quotation for the non-periodical evaluation is transferred to the second
examination period, which consists of a periodical examination only.

**Calculation of the examination mark**

Theory: 100% periodical evaluation
Exercises: Partly based on permanent evaluation (cf. the assignement mentioned
above). The permanent evaluation accounts for 25% of the final mark.
Students who are absent without any well-justified reason or who do not participate in
(part of) the permanent evaluation, do not pass the exam for this course unit.
The marks resulting from the permanent evaluation are retained in the second
examination period, as the second examination period only consists of a periodic
evaluation.